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Strontium Tetragermanate, SrGe₄O₉

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Abstract

The structure of tetragermanium strontium nonaoxide, SrGe₄O₉, comprises two different types of sheet structure stacked alternately along the *c* axis. One sheet comprises GeO₆ octahedra and SrO₈ polyhedra while the other sheet contains three-membered rings of GeO₄ tetrahedra. In addition, this crystal structure shows a basic unit of the superstructures which are found for PbGe₄O₉.

Comment

Three tetragermanates of formula *Me*Ge₄O₉ (*Me* = Sr, Pb, Ba) were studied by Robbins & Levin (1961). They showed that these compounds were isostructural on the basis of indexed powder X-ray diffraction patterns. In addition, they discussed a structural relationship with the mineral benitoite, BaTiSi₃O₉, the structure of which had been determined by Zachariasen (1930) and was later refined by Fischer (1969). Eulenberger, Wittmann & Nowotny (1962) synthesized two forms of CaGe₄O₉ (α and β types) and suggested that the α form was isostructural with the compounds studied by Robbins & Levin (1961).

Robbins, Perloff & Block (1966) determined the crystal structure of BaGe₄O₉ by single-crystal X-ray diffraction: *a* = 11.61, *c* = 4.74 Å, *P*3, *Z* = 3, final *R* value 6.8% for 327 reflections. Smolin (1969) also

studied the structure of the same compound. He reported unit-cell parameters that were consistent with those given by Robbins *et al.* (1966) but gave the space group as *P*321 with a final *R* value of 5.8% for 848 reflections.

Venevtsev *et al.* (1982) studied the structure and physical properties of the four polymorphs (α , β_1 , β_2 , γ) of PbGe₄O₉ and the solid solutions (Pb_{1-x}M_x)Ge₄O₉ (*M* = Sr, Ba). Both α - and γ -PbGe₄O₉ showed ferroelectric properties, while γ -PbGe₄O₉ also displayed ferroelastic properties. We have attempted to solve the crystal structure of SrGe₄O₉, which seems to be analogous to BaGe₄O₉, and to confirm which space group, *P*3 or *P*321, is correct.

As a result of the structure analysis, the space group of SrGe₄O₉ is *P*321, as reported by Smolin (1969) for BaGe₄O₉. The crystal structure of SrGe₄O₉ has two different types of sheet structure within it. One is formed by a combination of GeO₆ octahedra and SrO₈ polyhedra (Fig. 1) and the other by three-membered rings of GeO₄ tetrahedra (Fig. 2). The sheets are stacked alternately along the *c* axis (Fig. 3).

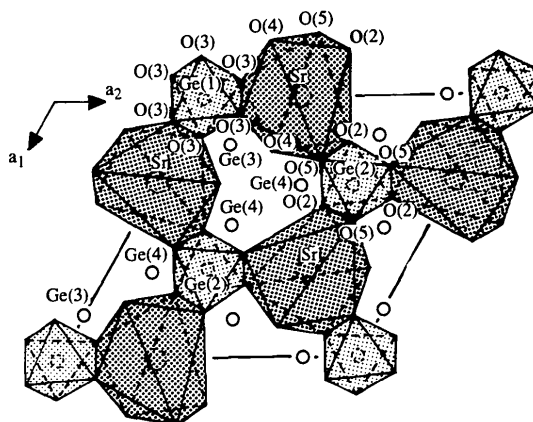


Fig. 1. The arrangement of the GeO₆ octahedra and the SrO₈ polyhedra viewed down the *c* axis.

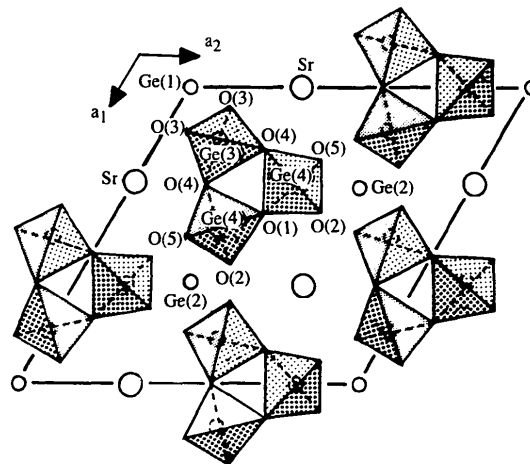


Fig. 2. The arrangement of the three-membered rings of GeO₄ tetrahedra viewed down the *c* axis.

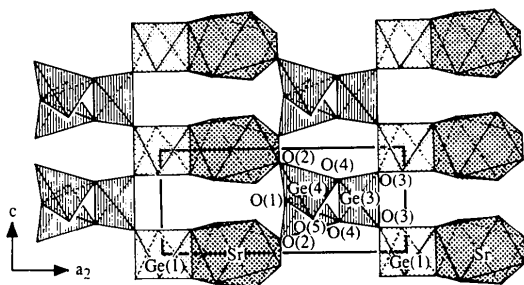


Fig. 3. The arrangement of the polyhedra about the cation sites with $|x| \leq 0.15$ projected on (100). Heavy lines indicate the unit cell.

The polymorphs of PbGe₄O₉ found by Venevtsev *et al.* (1982) are superstructures with different c periods. The lengths of their c axes are 4.75 (α phase), 9.49 (β_1 phase) and 19.09 Å (β_2 phase), suggesting that the α , β_1 and β_2 phases comprise a single unit, double units and four units of the basic structure, respectively. In conclusion, the structures of SrGe₄O₉ ($c = 4.75$ Å) and BaGe₄O₉ ($c = 4.74$ Å) are formed from the basic unit, and the structure of the mineral benitoite (BaTiSi₃O₉; $c = 9.76$ Å) is a superstructure with double units.

Experimental

Single crystals of strontium tetragermanate were synthesized from a stoichiometric mixture of SrCl₂ and GeO₂ at 1570 K. Electron microprobe analyses of the crystals showed that they had the chemical formula SrGe₄O₉.

Crystal data

SrGe₄O₉
 $M_r = 521.98$
 Trigonal
 P321
 $a = 11.344$ (2) Å
 $c = 4.750$ (2) Å
 $V = 529.4$ (3) Å³
 $Z = 3$
 $D_x = 4.91$ Mg m⁻³
 D_m not measured

Mo $K\alpha$ radiation
 $\lambda = 0.71073$ Å
 Cell parameters from 24 reflections
 $\theta = 19.8$ – 35.8°
 $\mu = 25.9$ mm⁻¹
 $T = 293$ K
 Sphere
 0.14 mm (radius)
 Colourless

Data collection

Rigaku AFC-5 diffractometer
 $\omega/2\theta$ scans
 Absorption correction: spherical
 $T_{\min} = 0.0113$, $T_{\max} = 0.0147$
 1452 measured reflections
 1343 independent reflections
 929 observed reflections
 $[I > 3\sigma(I)]$

$R_{\text{int}} = 0.025$
 $\theta_{\text{max}} = 40.0^\circ$
 $h = 0 \rightarrow 17$
 $k = 0 \rightarrow 17$
 $l = 0 \rightarrow 8$
 3 standard reflections monitored every 150 reflections
 intensity decay: 0.5%

Refinement

Refinement on F
 $R = 0.049$
 $wR = 0.052$
 $S = 1.35$
 929 reflections
 73 parameters
 Unit weights applied
 $(\Delta/\sigma)_{\text{max}} = 0.05$
 $\Delta\rho_{\text{max}} = 0.45$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.38$ e Å⁻³

Extinction correction: secondary isotropic
 Extinction coefficient: 4.40×10^{-5}
 Atomic scattering factors from *International Tables for X-ray Crystallography* (1974, Vol. IV)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å²)

$$B_{\text{eq}} = (4/3)\sum_i \sum_j \beta_{ij} a_i \cdot a_j$$

	x	y	z	B_{eq}
Sr	0.3280 (1)	0	0	1.40 (2)
Ge(1)	0	0	0	0.63 (3)
Ge(2)	1/3	2/3	0.1076 (4)	0.62 (3)
Ge(3)	0.8209 (1)	0	1/2	0.68 (4)
Ge(4)	0.4908 (1)	0.3410 (1)	0.3990 (2)	0.64 (3)
O(1)	0.5112 (9)	0	1/2	1.29 (8)
O(2)	0.6022 (8)	0.4222 (8)	0.1245 (14)	0.95 (7)
O(3)	0.1546 (7)	0.0616 (8)	0.7636 (13)	0.76 (7)
O(4)	0.3255 (8)	0.2145 (8)	0.2949 (16)	0.94 (8)
O(5)	0.5102 (7)	0.2496 (8)	0.6722 (13)	0.79 (7)

Table 2. Selected geometric parameters (Å, °)

Sr—O(2')	2.601 (7)	Ge(2)—O(5 ^{III})	1.860 (7)
Sr—O(2'')	2.601 (7)	Ge(2)—O(5 ^I)	1.860 (7)
Sr—O(3 ^{III})	2.644 (9)	Ge(2)—O(5 ^{II})	1.860 (7)
Sr—O(3 ^{II})	2.644 (9)	Ge(2)—O(2 ^{III})	1.872 (9)
Sr—O(4)	2.820 (9)	Ge(2)—O(2 ^{III})	1.872 (9)
Sr—O(4')	2.820 (9)	Ge(2)—O(2 ^{II})	1.872 (9)
Sr—O(5 ^{III})	2.977 (8)	Ge(3)—O(3')	1.747 (6)
Sr—O(5 ^I)	2.977 (8)	Ge(3)—O(3 ^{II})	1.747 (6)
Ge(1)—O(3 ^{III})	1.897 (7)	Ge(3)—O(4')	1.789 (7)
Ge(1)—O(3 ^{II})	1.897 (7)	Ge(3)—O(4 ^{II})	1.789 (7)
Ge(1)—O(3 ^I)	1.897 (7)	Ge(4)—O(2)	1.726 (7)
Ge(1)—O(3 ^{III})	1.897 (7)	Ge(4)—O(5)	1.743 (8)
Ge(1)—O(3 ^{II})	1.897 (7)	Ge(4)—O(1 ^{III})	1.755 (12)
Ge(1)—O(3 ^I)	1.897 (7)	Ge(4)—O(4)	1.769 (7)
Ge(4')—O(1)—Ge(4 ^{II})	123.8 (5)	Ge(3 ^{III})—O(4)—Ge(4)	121.2 (4)

Symmetry codes: (i) $1 - y, x - y, z$; (ii) $1 - x, y - x, -z$; (iii) $x, y, z - 1$; (iv) $x - y, -y, 1 - z$; (v) $x - y, -y, -z$; (vi) $-y, x - y, z - 1$; (vii) $y - x, -x, z - 1$; (viii) $y, x, 1 - z$; (ix) $-x, y - x, 1 - z$; (x) $1 - x, 1 + y - x, 1 - z$; (xi) $x - y, 1 - y, 1 - z$; (xii) $y, x, -z$; (xiii) $1 - x, 1 + y - x, -z$; (xiv) $x - y, 1 - y, -z$; (xv) $1 - x, y - x, 1 - z$; (xvi) $1 + y - x, 1 - x, z$.

The intensities of the reflections were measured at the X-ray laboratory of Nihon University, Japan.

Data collection, cell refinement and data reduction: *AFC/MSD Diffractometer Control Program* (Rigaku Corporation, 1991). Program used to solve and refine structure: *LINUS* (Coppens & Hamilton, 1970).

Lists of structure factors, anisotropic displacement parameters and complete geometry have been deposited with the IUCr (Reference: OH1096). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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